

Infosafe No™ VARH9	Issue Date : March 2016	ISSUED by HUNTER
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 Product Name **OPL COMMERCIAL BLEACH**

Classified as hazardous

## 1. Identification

<b>GHS Product Identifier</b>	OPL COMMERCIAL BLEACH
<b>Company Name</b>	Hunter Industrials Pty Ltd (ABN 083 330 974)
<b>Address</b>	51-53 Lakewood Blvd Braeside Vic 3195 Australia
<b>Telephone/Fax Number</b>	Tel: (03) 9586 2888 Fax: (03) 9587 9851
<b>Emergency phone number</b>	0409 949 298
<b>E-mail Address</b>	sales@huntind.com.au
<b>Recommended use of the chemical and restrictions on use</b>	Chlorine bleach for use in commercial laundries.

## 2. Hazard Identification

<b>GHS classification of the substance/mixture</b>	Corrosive to Metals: Category 1 Eye Damage/Irritation: Category 1 Skin Corrosion/Irritation: Category 1B
<b>Signal Word (s)</b>	DANGER
<b>Hazard Statement (s)</b>	EUH031 Contact with acids liberates toxic gas. H290 May be corrosive to metals. H314 Causes severe skin burns and eye damage. H318 Causes serious eye damage.
<b>Precautionary statement – General</b>	P102 Keep out of reach of children. P103 Read label before use.
<b>Pictogram (s)</b>	Corrosion  
<b>Precautionary statement – Prevention</b>	P234 Keep only in original container. P260 Do not breathe dust/fume/gas/mist/vapours/spray. P264 Wash contaminated skin thoroughly after handling P280 Wear protective gloves/protective clothing/eye protection/face protection.
<b>Precautionary statement – Response</b>	P301+P330+P331 IF SWALLOWED: rinse mouth. Do NOT induce vomiting. P303+P361+P353 IF ON SKIN (or hair): Remove/Take off immediately all contaminated clothing. Rinse skin with water/shower. P304+P340 IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing. P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. P310 Immediately call a POISON CENTER or doctor/physician. P363 Wash contaminated clothing before reuse. P390 Absorb spillage to prevent material damage.
<b>Precautionary statement – Storage</b>	P405 Store locked up.
<b>Precautionary statement – Disposal</b>	P501 Dispose of contents/container to authorised chemical landfill or if organic to high temperature incineration
<b>Other Information</b>	LD 50 : No data LC 50 : Chlorine 293 ppm/1 hour rat LCLo : Chlorine 2,530 mg/m3/30 minutes human

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### 3. Composition/information on ingredients

Ingredients	Name	CAS	Proportion
	Sodium Hypochlorite 12.5%	7681-52-9	100%

### 4. First-aid measures

<b>Inhalation</b>	Remove from exposure, rest and keep warm. In severe cases, obtain medical attention. Apply artificial respiration if not breathing.
<b>Ingestion</b>	Immediately rinse mouth with water. Do NOT induce vomiting. Slowly give water to drink. Seek medical assistance.
<b>Skin</b>	If skin contact occurs, remove contaminated clothing and wash skin thoroughly. If swelling, redness, blistering, or irritation occurs seek medical advice.
<b>Eye contact</b>	If in eyes, hold eyes open, flood with water for at least 15 minutes and see a doctor.
<b>First Aid Facilities</b>	Eye wash station and normal washroom facilities.
<b>Advice to Doctor</b>	Product is a solution of sodium hypochlorite. If swallowed, may lead to fall in blood pressure. Treat with antacids to neutralise hypochlorous acid formed in the stomach, then as for alkaline materials. Onset of pulmonary oedema, following inhalation overexposure, may be delayed. Treat symptomatically. Contact Poisons Information Centre.

### 5. Fire-fighting measures

<b>Suitable extinguishing media</b>	Use extinguishing media appropriate to surrounding fire.
<b>Hazards from Combustion Products</b>	Chlorine, water vapour, sodium hydroxide, sodium carbonate, sodium chloride.
<b>Specific Methods</b>	In case of small fire/explosion use water. In case of major emergency use PPE: breathing apparatus and protective gloves.
<b>Specific hazards arising from the chemical</b>	Contact with combustible material may cause fire. May form explosive products with primary aliphatic or aromatic amines, methanol and with nitrites. Contact with acids will generate chlorine, a toxic and corrosive gas. May react vigorously or violently with oxidising agents, reducing agents and metal salts.
<b>Hazchem Code</b>	2X
<b>Other Information</b>	Avoid contact with coloured fabric as Chlorine may bleach colour out. May give off dangerous gas if mixed with other products.

### 6. Accidental release measures

<b>Spills &amp; Disposal</b>	Disposal of small spillages only. For large spillages liquids should be contained using sand or earth, and both liquids and solids then transferred to salvage containers. Residues should be treated as for small spillages. CAUTION: Before dealing with spillage take necessary protective measures, inform others to keep at a safe distance and, for flammable materials, shut off all possible sources of ignition. If local regulations permit, mop up with plenty of water and run to waste, diluting greatly with running water. Otherwise absorb on inert absorbent, transfer to sealed container and arrange removal by disposals company. Wash site of spillage thoroughly with water and detergent. Ventilate area to dispel any residual vapours.
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### 7. Handling and storage

<b>Conditions for safe storage, including any incompatibilities</b>	Store in a cool, well ventilated place, out of reach of children. Large quantities should be stored in a bunded dangerous goods store. Store in original container. Never store in unlined metal containers. Keep container tightly closed. Keep out of direct sunlight. Keep away from combustible materials. Keep away from acids. Keep away from metals and their salts. Keep away from aliphatic and aromatic amines. Keep away from methanol and nitrites. Keep away from oxidising and reducing agents. Prevent vapours
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from collecting in enclosed spaces. Protect from physical damage. Clean up all spills and splashes promptly; avoid secondary accidents.

## 8. Exposure controls/personal protection

<b>Appropriate engineering controls</b>	Prevent direct contact with metals. Local mechanical exhaust/extraction usually required to keep airborne contamination as low as possible.
<b>Personal Protective Equipment</b>	Prevent contact with the eyes. Avoid contact with the skin. Avoid breathing the vapours. Personal protection to be selected from those recommended below, as appropriate to mode of use, quantity handled and degree of hazard:- Self contained breathing apparatus Face shield, goggles or safety glasses Gloves, rubber or plastic Plastic apron, sleeves and boots Impervious overalls. Always maintain a high level of personal hygiene when using cleaning chemicals. That is wash hands before eating, drinking, smoking or using the toilet.

## 9. Physical and chemical properties

<b>Form</b>	Liquid
<b>Appearance</b>	Clear yellow liquid
<b>Colour</b>	Clear pale yellow.
<b>Odour</b>	Strong smell of chlorine.
<b>Boiling Point</b>	100C
<b>Solubility in Water</b>	Miscible with water in all proportions.
<b>Specific Gravity</b>	1.19
<b>pH</b>	12.0 - 13.0
<b>Vapour Pressure</b>	Not available.
<b>Flash Point</b>	>61C
<b>Flammability</b>	Not flammable. Moderate oxidiser. Contact with combustible materials may cause fire.
<b>Other Information</b>	Oxidiser. Contact with combustible material may cause fire. Contact with acids will generate chlorine, a toxic and corrosive gas. May react violently with reducing agents. Can react with primary aliphatic and aromatic amines, methanol and nitrites to give explosive products. May react vigorously with oxidising agents. Incompatible with most metals. Will decompose on standing, generating chlorine. Decomposition will be accelerated by contamination and by exposure to sun light. May react vigorously with peroxides and metal salts. On long storage, may generate pressure inside sealed containers. Open cautiously.

## 10. Stability and reactivity

<b>Conditions to Avoid</b>	Incompatible materials, heating, sunlight, prolonged exposure to air.
<b>Incompatible Materials</b>	Acids, other oxidising agents, combustible materials, amines, ammonium compounds, formic acid, methanol, phenylacetonitrile, reducing agents, metal salts, metals, wood and wood products.
<b>Hazardous Decomposition Products</b>	Decomposes on heating to emit toxic fumes. Heating can cause expansion of containers or decomposition leading to violent rupture of containers. Reacts vigorously with acids to produce dangerous levels of gaseous chlorine. Reacts with amines, ammonium salts, aziridine, methanol, phenylacetonitrile, metal salts, peroxides and reducing agents.
<b>Possibility of hazardous reactions</b>	My form toxic oxides of Chlorine if involved in a fire.

## 11. Toxicological Information

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<b>Acute Toxicity - Oral</b>	Acute oral toxicity (LD50): 5800 mg/kg [Mouse]
<b>Acute Toxicity - Inhalation</b>	LCLo : Chlorine 2,530 mg/m3/30 minutes human
<b>Ingestion</b>	Will cause severe irritation and corrosion of the mouth, throat and gastrointestinal system. May cause pain and vomiting. May cause fall in blood pressure. Risk of delirium, coma and death.
<b>Inhalation</b>	Inhalation of chlorine gas at 1 ppm will irritate the mouth, nose and throat. Above 1.3 ppm, vapours may cause coughing and difficulty breathing. At higher levels, risk of throat muscle spasm, leading to death by suffocation. Risk of delayed onset of pulmonary oedema (fluid in the lungs).
<b>Skin</b>	Short contact may cause irritation. On longer contact risk of chemical burns.
<b>Eye</b>	Severe irritant. Risk of permanent damage and blindness.
<b>Chronic Effects</b>	Repeated skin contact may lead to dermatitis or 'chloracne'. Repeated, low level exposure to chlorine vapours may cause corrosion of the teeth.

## 12. Ecological information

<b>Ecotoxicity</b>	Toxic to fish and aquatic organisms.
<b>Persistence and degradability</b>	Will degrade on exposure to air, sunlight.
<b>Mobility</b>	Readily transported by water and through soil.
<b>Environmental Fate</b>	Decomposes over time.
<b>Environmental Protection</b>	Avoid contaminating waterways, drains, sewers, or ground.

## 13. Disposal considerations

<b>Waste Disposal</b>	Do not discharge effluent containing this product into laes, streams, ponds, estuaries, oceans, or public waters unless this product is specifically identified and addressed in a permit. Do not discharge effluent containing this product without previously notifying the sewage treatment plant authority. For guidance, contact your State Water Board or Regional Office of the EPA. Reduce with agents such as bisulfites or ferrous salt solutions. Some heat will be produced. Keep on alkaline side and dilute with copious amounts of water. The main end-product is salt water. Waste must be disposed of in accordance with federal, state and local environmental control regulations. Refer to Land Waste Management Authority in your State.
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## 14. Transport information

<b>Transport Information</b>	Classified as a Class 8 Dangerous Good. Dangerous Goods of Class 8 Corrosives are incompatible in a placard load with any of the following: - Class 1, Class 4.3, Class 5, Class 6, if the Class 6 dangerous goods are cyanides and the Class 8 dangerous goods are acids and Class 7. Store away from acids.
<b>U.N. Number</b>	1791
<b>UN proper shipping name</b>	HYPOCHLORITE SOLUTION
<b>Transport hazard class(es)</b>	8
<b>Hazchem Code</b>	2X
<b>Packaging Method</b>	3.8.8RT7,RT8
<b>Packing Group</b>	III
<b>EPG Number</b>	8A1
<b>IERG Number</b>	37

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**15. Regulatory information**

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<b>Poisons Schedule</b>	S5
<b>Hazard Category</b>	Corrosive
<b>AICS (Australia)</b>	All components listed.

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**16. Other Information**

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<b>Signature of Preparer/Data Service</b>	Technical Manager 03 9580 2499
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<b>Technical Contact Numbers</b>	Emergency Advice All Hours: Tel: 0409 949 298 Mon-Fri 8am - 6pm Poisons Information Centre: 13 11 26 - 24hrs
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<b>Other Information</b>	This SDS summarises at the date of issue our best knowledge of the health and safety hazard information of the product, and in particular how to safely handle and use the product in the Workplace. Please refer to the technical datasheet (Instructions for use), and the label on the drum. The company cannot anticipate or control the individual working conditions encountered and so each user should read this SDS carefully, and if in doubt ring the Contact Point Number given below. ...End Of MSDS...
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